

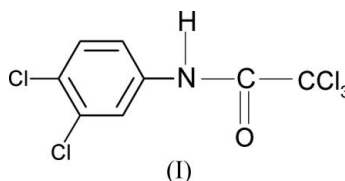
2,2,2-Trichloro-*N*-(3,4-dichlorophenyl)acetamideB. Thimme Gowda,^{a*} Jozef Kožíšek,^b Miroslav Tokarcík^c and Hartmut Fuess^d^aDepartment of Chemistry, Mangalore University, Mangalagangotri 574 199, Mangalore, India, ^bDepartment of Physical Chemistry, Slovak University of Technology, Radlinského 9, SK-812 37 Bratislava, Slovak Republic, ^cDepartment of Chemical Physics, Slovak University of Technology, Radlinského 9, 812 37 Bratislava, Slovak Republic, and ^dInstitute of Materials Science, Darmstadt University of Technology, Petersenstrasse 23, D-64287 Darmstadt, Germany

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Key indicators

Single-crystal X-ray study
 $T = 295$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
Disorder in main residue
 R factor = 0.028
 wR factor = 0.067
Data-to-parameter ratio = 14.1For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.In the title compound, $\text{C}_8\text{H}_4\text{Cl}_5\text{NO}$, the N–H bond is *anti* to the *meta*-Cl substituent, similar to the chain-unsubstituted *N*-(3,4-dichlorophenyl)acetamide, although the two amides crystallize in different crystal systems.

Comment

In the present work, the structure of 2,2,2-trichloro-*N*-(3,4-dichlorophenyl)acetamide (34DCPTCA), (I), has been determined to explore the substituent effects on the structure of *N*-aromatic amides (Gowda *et al.*, 2006, 2007). The substitution of Cl for H atoms in the side-chain methyl group of *N*-(3,4-dichlorophenyl)acetamide (34DCPA) changes the crystal system from triclinic to monoclinic (Jones *et al.*, 1990), but the conformation of the N–H bond continues to be *anti* to the *meta*-Cl substituent (Fig. 1). The substitution in the side chain increases the bond lengths of C4–C5 and C8–N7 from 1.503 (4) to 1.549 (2) Å and 1.409 (3) to 1.424 (2) Å, respectively, but decreases the lengths of C5–O6 and C5–N7 from 1.221 (3) to 1.209 (2) Å and 1.351 (3) to 1.336 (2) Å, respectively. The C5–N7–C8 angle changes from 128.2 (2) to 123.6 (1)° (Table 1). The packing diagram of the title compound, as viewed down the *a* axis, is shown in Fig. 2.

Experimental

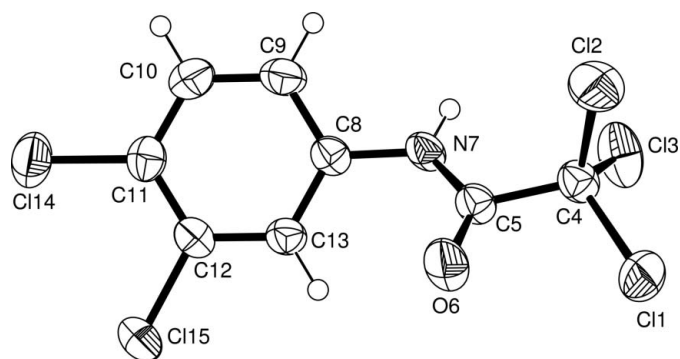
The title compound was prepared according to the literature method (Shilpa & Gowda, 2007). Single crystals of the title compound were obtained by recrystallization from an ethanol solution.

Crystal data

$\text{C}_8\text{H}_4\text{Cl}_5\text{NO}$	$V = 1148.33$ (6) Å ³
$M_r = 307.37$	$Z = 4$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 5.8610$ (2) Å	$\mu = 1.23$ mm ⁻¹
$b = 20.2384$ (6) Å	$T = 295$ (2) K
$c = 9.9791$ (3) Å	$0.28 \times 0.14 \times 0.11$ mm
$\beta = 104.041$ (3)°	

Data collection

Oxford Diffraction Xcalibur diffractometer	15232 measured reflections
Absorption correction: multi-scan (<i>CrysAlis RED</i> ; Oxford Diffraction, 2006)	2357 independent reflections
$T_{\min} = 0.816$, $T_{\max} = 0.877$	2045 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.018$


Figure 1

The molecular structure of the title compound, showing the atom labelling scheme. Displacement ellipsoids are drawn at the 50% probability level.

Refinement

$$R[F^2 > 2\sigma(F^2)] = 0.028$$

$$wR(F^2) = 0.067$$

$$S = 1.05$$

2357 reflections

167 parameters

4 restraints

H-atom parameters constrained

$$\Delta\rho_{\max} = 0.31 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.34 \text{ e } \text{\AA}^{-3}$$

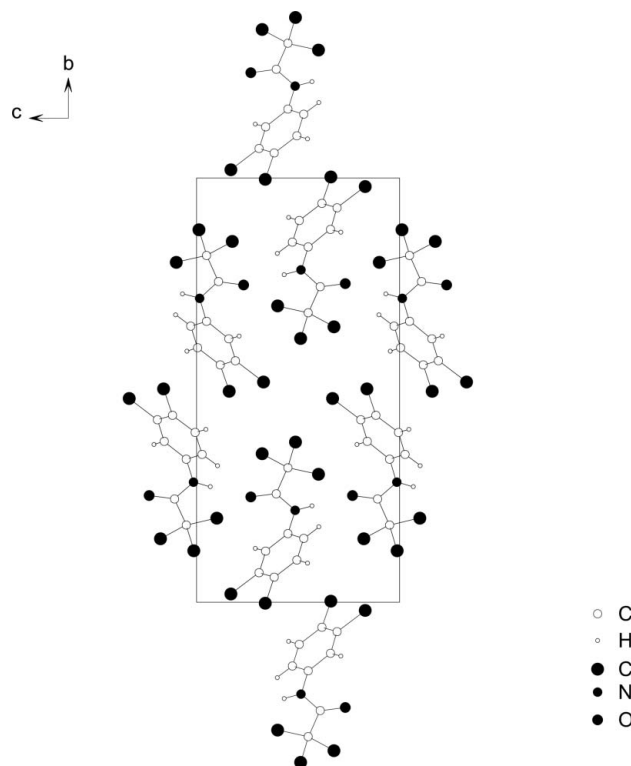
Table 1

Selected geometric parameters (\AA , $^\circ$).

C4—C5	1.549 (2)	C5—N7	1.336 (2)
C5—O6	1.2092 (18)	C8—N7	1.424 (2)
O6—C5—N7	124.71 (15)	C13—C8—N7	121.37 (14)
O6—C5—C4	119.15 (14)	C9—C8—N7	118.84 (14)
N7—C5—C4	116.05 (13)	C5—N7—C8	123.60 (13)
C13—C8—C9	119.79 (15)		
N7—C8—C9—C10	-178.49 (16)	C13—C8—N7—C5	-40.9 (2)
N7—C8—C13—C12	178.36 (14)	C9—C8—N7—C5	139.49 (17)
C4—C5—N7—C8	-172.72 (14)		

C-bound H atoms were refined using a riding model with C—H distances of 0.93 \AA . A restraint of N—H = 0.86 (2) \AA was applied for the N-bound H atom. For all H atoms $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$. The title compound exhibits positional disorder of atoms Cl11, Cl12 and Cl13 bonded to atom C4, with the two disorder components rotated relative to each other by about 14° . The C—Cl bond lengths were restrained to 1.76 (1) \AA . The refined occupancy factors are 0.56:0.44 (3).

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2006); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2006); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows*


Figure 2

The packing, as viewed down the *a* axis.

(Farrugia, 1997); software used to prepare material for publication: *SHELXL97*, *PLATON* (Spek, 2003) and *WinGX* (Farrugia, 1999).

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References

- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
 Gowda, B. T., Kozisek, J. & Fuess, H. (2006). *Z. Naturforsch. Teil A*, **61**, 588–594.
 Gowda, B. T., Kozisek, J., Svoboda, I. & Fuess, H. (2007). *Z. Naturforsch. Teil A*, **62**, 91–100.
 Jones, P. G., Kirby, A. J. & Lewis, R. J. (1990). *Acta Cryst.* **C46**, 78–81.
 Oxford Diffraction (2006). *CrysAlis CCD* and *CrysAlis RED*. Oxford Diffraction Ltd, Abingdon, Oxfordshire, England.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
 Shilpa & Gowda, B. T. (2007). *Z. Naturforsch. Teil A*, **62**, 84–90.
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.